

Mechanism for increasing dopant incorporation in semiconductors via doped nanostructuresIgor L. Kuskovsky,^{1,2,*} Y. Gu,^{2,†} Y. Gong,² H. F. Yan,² J. Lau,^{2,4} I. C. Noyan,² G. F. Neumark,² O. Maksimov,³
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A long-standing problem for ZnSe (and related alloys) has been to obtain good p -type doping. Recent work has given about an order-of-magnitude improvement in such doping by use of Te as a “codopant” to facilitate the introduction of an acceptor dopant (N), since it is known that p -ZnTe can be obtained quite readily; the Te was introduced in submonolayer quantities via planar (δ) doping during molecular beam epitaxy. Here, we examine the mechanism of this improved doping. We show that it resides in the formation of ZnTe-rich nanoislands, with the N embedded in these. This result is obtained by studies involving transmission electron microscopy, high-resolution x-ray diffraction, secondary-ion mass spectroscopy, and temperature quenching of photoluminescence. We note that these nanoislands appear quite unique, in providing doping of semiconductors, and thus are of great interest of their own.

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It is well known that it has been difficult to obtain “adequate” (high) bipolar conductivity in wide-band-gap semiconductors, and that this problem resides in obtaining adequate dopant incorporation due to poor solubility and/or excessive compensation (e.g. Refs. 1–4). It is also well known that in some wide-band-gap semiconductors it is easy to incorporate n -type dopants, whereas in others p -type doping is easy; the cause for this behavior is of great physical interest, but microscopic models (e.g., Refs. 5–8 and references therein) are still limited in scope or not yet well established. In any case, we have utilized^{9,10} this type of behavior to obtain about an order-of-magnitude improvement in dopant incorporation in p -type Zn(Be)Se, a material which is notoriously difficult to dope p type. Specifically, we grew the Zn(Be)Se via molecular beam epitaxy (MBE) and incorporated δ -layers containing N as acceptor dopant and Te as “codopant” (i.e., material used to facilitate introduction of the dopant); Te was used since it is known that p -ZnTe can be readily obtained (e.g., Ref. 11 and references therein). Two types of samples were grown, one with insertion of a single δ -layer (δ -doped samples), the other with insertion of three contiguous δ -layers (δ^3 -doped samples) between nominally undoped ZnSe layers, which heretofore we shall refer to as spacers; we note that it was the latter samples that showed the most improved doping.^{9,10} We further note that very small amounts of Te were used, resulting in a coverage of ~ 0.25 monolayer (ML) (here 1 ML is half of the unit cell) as obtained from high-resolution x-ray diffraction (HRXRD) (see below).

In the present paper, we present evidence that the microscopic mechanism leading to this improved doping (in the δ^3 -doped samples) resides in N being embedded in nanoislands rich in ZnTe.¹² Moreover, it also must be emphasized that these Te-N nanostructures are of a unique type (to the best of our knowledge) having the property that they provide

doping in semiconductors, and thus are of very great interest on their own. The evidence for the ZnTe:N nanoislands is given by (i) transmission electron microscopy (TEM), which showed that the δ -regions do not form continuous planes; thus, there must be formation of islands, which, as shown below, are nanosized; (ii) HRXRD which showed that δ -regions are rich in ZnTe, i.e., the nanoislands are Te rich; (iii) secondary-ion mass spectroscopy (SIMS) in conjunction with electrochemical capacitance-voltage (ECV) profiling, which showed that both N and Te stay together within the δ -region and that N does not form compensating complexes, as it does in bulk ZnSe:N (see, e.g., Refs. 13–16 and references therein); (iv) the temperature quenching of photoluminescence (PL), from which we show that the resultant values of the N acceptor activation energies can be explained only by assuming that N is embedded preferentially in a ZnTe-rich environment.

Various sample configurations based on the basic sample structure shown in Fig. 1(a) were designed for a specific characterization technique (TEM, HRXRD, and SIMS) to achieve the optimal results. For HRXRD, “standard”⁹ samples were grown with a Te+N deposition time of 5 s and with 10 ML ZnSe (nominally undoped) spacer regions; for TEM, a sample with a 20 s Te+N deposition time [which is longer than the standard time, in order to enhance the TEM contrast associated with the substitution of Se by Te (Ref. 17)] and 6 ML spacers (nominally undoped ZnSe) was prepared; for SIMS, a “supersample,” consisting of four basic sample structures (with 48 ML spacers) obtained by changing the (Te+N) deposition time from 5 to 40 s [regions A, B, C, and D in Fig. 1(b)], separated by 300-nm-thick ZnSe buffers, was grown.

We first discuss the TEM results. Bright- dark-field imaging, high-resolution electron microscopy (HREM), and selective area electron diffraction (SAED) obtained from a

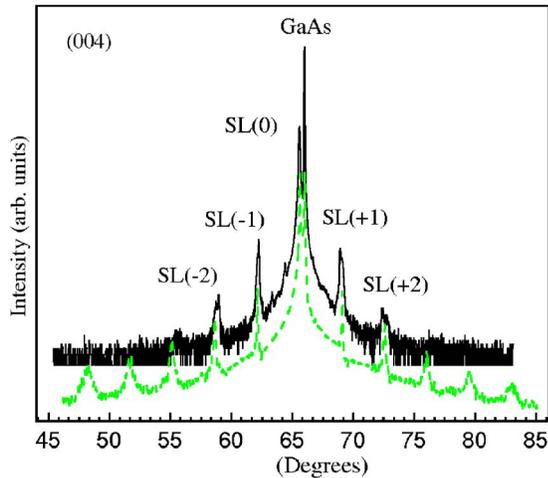


FIG. 3. (Color online) The (004) θ - 2θ scan [solid (black) line] of a sample with ~ 10 ML thick spacers and 5 s Te+N deposition time. The dashed (green) line is the best-fit simulation, using dynamical diffraction theory.

are in excellent agreement with the nominal growth parameters; the average Te concentrations are 37% and 2.2% in the δ -layers and the spacers, respectively. Moreover, a broadening of the satellite peaks was observed; this was modeled during the simulations by a period dispersion (e.g., Ref. 21 and references therein). Furthermore, the low coverage and relatively high Te concentration within the δ^3 -layers indicate that Te is not uniformly distributed within these layers, and, thus, forms ZnTe-rich nanoislands (ZnTe and ZnSe type-II quantum dots have been observed optically in similar samples grown without nitrogen²²). The high Te concentration in the δ -layers is consistent with doping results obtained for ZnSeTe alloys, where it was shown that high acceptor concentrations are observed only for Te concentrations exceeding 15%.¹¹ The incorporation of minute quantities of Te into the nominally undoped spacer regions is most likely due to the Te presence in the molecular beam epitaxy chamber even after the Te shutters are closed.

Having established the existence of a periodic structure of these nanoislands, we next discuss results of SIMS measurements in combination with ECV depth profiling.²³ First we show SIMS (Fig. 4) on a standard sample together with the depth profile of the net acceptor concentration ($[N_A - N_D]$) (N_A and N_D are the total acceptor and donor concentrations, respectively) obtained via ECV profiling obtained previously (see Ref. 9). This gives a direct comparison between $[N_A - N_D]$ and the total nitrogen concentration. The average $[N]$ in the film (δ -layers and spacers) is only slightly higher than that in the ZnSe:N buffer, which is probably due to an increased solubility in the presence of Te.²⁴ On the other hand, $[N_A - N_D]$ in the film and buffer are dramatically different: in the film, $[N_A - N_D] \approx [N]$, whereas in the buffer layer $[N_A - N_D]$ is only one-tenth of $[N]$. Moreover, $[N_A - N_D]$ decreases only when the concentration of Te decreases drastically. This low $[N_A - N_D]$ is consistent with extensive evidence that a large portion of nitrogen in bulk ZnSe:N is incorporated on “compensating” sites (e.g., Refs. 13–19). It is of great importance that $[N_A - N_D]$ in the film with

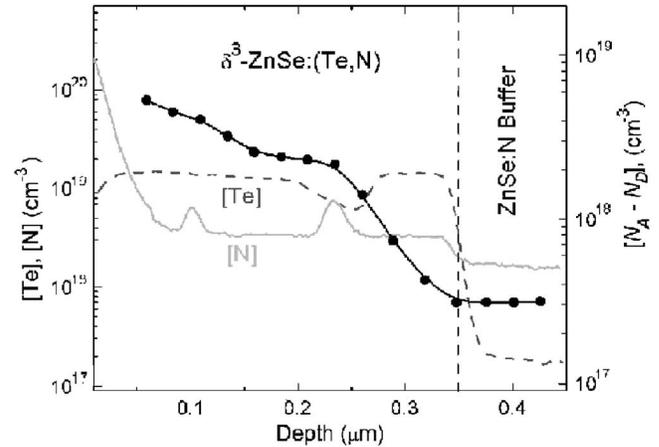


FIG. 4. SIMS and ECV profiles for a sample with ~ 10 ML thick spacers and 5 s Te+N deposition time. Left scale: dashed line is the total Te concentration, solid line is the total N concentration. Right scale: solid circles are net acceptor concentration. The vertical dashed line represents the border between the film (δ layers and spacers) and the ZnSe:N buffer layer.

δ -regions is far higher, and we thus conclude that in the δ^3 -doped ZnSe:(Te,N) nitrogen stays preferentially within the ZnTe environment, where it acts as an acceptor.

To further confirm that Te and N are located spatially preferentially within the same regions, we performed SIMS on the supersample,²⁵ which had thicker spacers so that one could identify the position of N and Te within the δ -layers. Indeed, the periodic, synchronous variations in the Te and the N compositions are clearly seen in Fig. 5, especially in the top portion of the supersample, where the SIMS resolution is the highest.²⁶ These variations occur on the scale of 10–11 nm, which is in good agreement with the nominal spacer thickness (48 ML). In other words, in our current system, even given the low Te coverage, the Te “locks” the nitrogen in place, preventing its diffusion into the ZnSe spac-

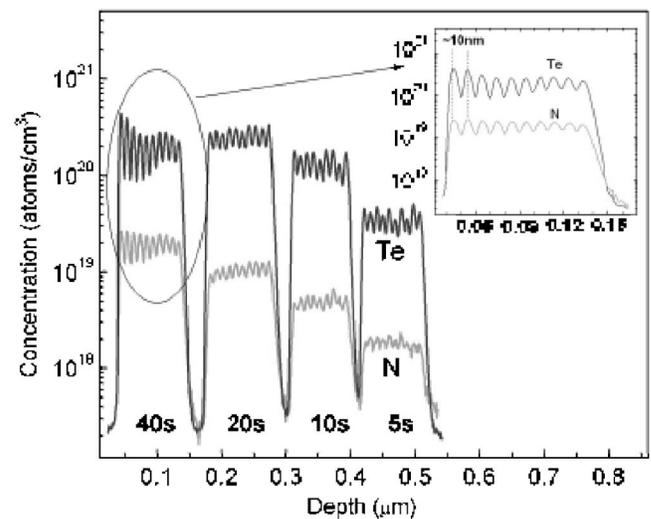


FIG. 5. SIMS profile for the supersample [schematics shown in Fig. 1(b)] showing concurrent periodic variations in Te and N concentrations on the scale of ~ 10 nm.

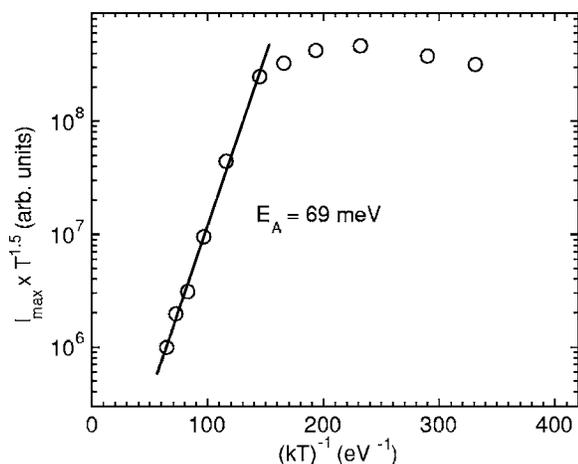


FIG. 6. PL of one typical sample (circles) at various temperatures. Solid line is from fitting.

ers, where it would behave as a donor, increasing compensation in the ZnSe:N, and causing devices failures.^{13,15}

Still further, independent, evidence for nitrogen incorporation into Te-rich nanoislands is given by temperature quenching of the PL (Fig. 6). For the detailed description of PL see Ref. 27. The resultant activation energy of N in δ^3 -doped samples is appreciably lower (see Table I) than that in ZnSe, being 38–87 meV,²⁸ vs 111 meV in ZnSe;²⁹ we note that the N activation energy in ZnTe has been reported as 30–65 meV,^{30–32} i.e., our values are close to those in ZnTe. These results show that the N must be located in a nanoisland, which contains a relatively large fraction of Te as next-nearest neighbors (Zn atoms are the nearest neighbors). Moreover, this activation energy decreases with increasing Te concentration (Table I). This is as expected: a higher Te concentration is expected to result in a larger percentage of Te (vs Se) surrounding a N atom, and thus lead to an activation energy closer to the ZnTe value.

We further would like to point out two important practical aspects, which arise from the above discussions. First, since the acceptors are now not located within ZnSe, but rather in a “favorable” ZnTe-rich environment, one can attempt to use other dopants (e.g., As, P, or Sb) which give high carrier concentrations in ZnTe, but have not succeeded in ZnSe. In

TABLE I. Sample parameters and PL properties.

Te concentration	$N_A - N_D$ (cm^{-3})	PL quenching activation energy (meV)
<3.0%	$\sim 6.0 \times 10^{18}$	38
$\sim 1.3\%$	$\sim 4.0 \times 10^{18}$	72
<1.0%	$\sim 3.0 \times 10^{18}$	87

addition, we again note, and this is quite important, that there is strong evidence that nitrogen itself could be a culprit in device degradation.^{13,15} Second, a similar approach could be applied to achieving adequate *p*-type doping of other wide-band-gap semiconductors, including ZnO and $\text{Al}_x\text{Ga}_{1-x}\text{N}$ with $x \geq 0.15$, materials currently of high physical and technological importance, for which, however, good *p*-type doping is still difficult to attain.³³ It appears that if one uses co- δ -doping with As, it would be equivalent to the above ZnSeTe system. In fact, GaAs can be *p*-type doped relatively easily using Mg, C, or Be (we note that the C site is equivalent to the N site in ZnSe); moreover, these are smaller than As, which in turn is larger than N, so the same driving forces for nanoisland formation are expected.

In summary, we have shown the existence of N-doped ZnTe-rich nanoislands in δ^3 -ZnSe:(Te,N) multilayers, grown by MBE, using N as a dopant and Te as a codopant to facilitate the incorporation of N. These nanoislands are certainly unusual, and may well be unique, providing vastly improved doping in at least one semiconductor system. The proof for these islands is provided by TEM, which showed a periodic structure of sub-monolayer islands; by HRXRD, which showed the islands to be Te-rich; by SIMS in conjunction with ECV profiling, which showed that both N and Te stay together within the δ^3 -regions; and by the temperature quenching of PL, which showed that the resultant values of the N acceptor activation energies are consistent with N being embedded preferentially in a ZnTe-rich environment.

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